

# In-Space Manufacture of Storable Propellants

Completed Technology Project (2015 - 2016)



## Project Introduction

Many deep-space, missions, especially those that return material or crews to near-Earth space, are severely limited by the need to carry propellants and heat shields to achieve their mission goals. Lifting these assets from the surface of Earth, landing them on the target body, launching them from there into an Earth-intercept trajectory, and capturing them into Earth orbit requires Earth launch of masses of propellant that increase exponentially with the mission's total delta V requirement. Preliminary studies of the logistics of gathering material from the Moon and selected Near-Earth Asteroids (NEAs) have demonstrated very large enhancements of mass-retrieval capabilities using propellants derived from sources in space rather than propellants launched from Earth and carried throughout the mission. They also have clearly shown the enormous advantages inherent in deriving propellants from NEAs. This study examines water-based propulsion using NEA volatiles to manufacture storable chemical propellants. The problem of storable propellants on Earth has been solved by the use of hydrazine derivatives as fuel and  $\text{N}_2\text{O}_4$  as oxidizer, both made possible by Earth's nitrogen-rich atmosphere. Nitrogen is scarce on asteroids, and would be best devoted to creating fire-retardant atmospheres for crews. There are plausible paths known for making asteroid-derived carbon-based storable fuels, but the provenance of a suitable storable oxidizing agent that does not employ nitrogen is an unsolved and difficult problem.

## Anticipated Benefits

Enormous improvement in mass-retrieval capabilities from asteroids and other bodies: mass payback of order 100:1 (retrieved payload/mass launched from LEO) instead of 0.1:1 or less for conventional Earth-based chemical propulsion. This is a true game-changer that opens the inner Solar System to efficient exploration and commercial exploitation, vastly increasing the mass of resources available to support human activities on Earth and in space.



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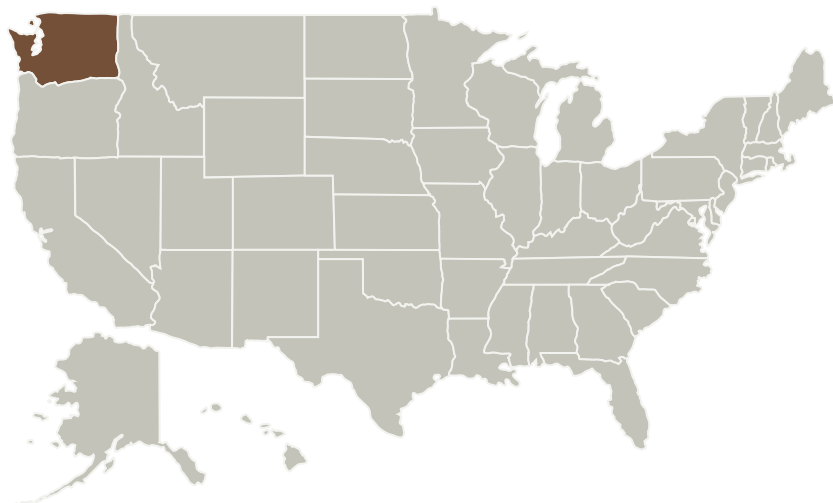
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## Primary U.S. Work Locations and Key Partners




Organizations Performing Work	Role	Type	Location
Deep Space Industries, Inc.	Lead Organization	Industry	San Jose, California

## Primary U.S. Work Locations

Washington

## Project Transitions

 **July 2015:** Project Start

## Organizational Responsibility

**Responsible Mission Directorate:**

Space Technology Mission Directorate (STMD)

**Lead Organization:**

Deep Space Industries, Inc.

**Responsible Program:**

NASA Innovative Advanced Concepts

## Project Management

**Program Director:**

Jason E Derleth

**Program Manager:**

Eric A Eberly

**Principal Investigator:**

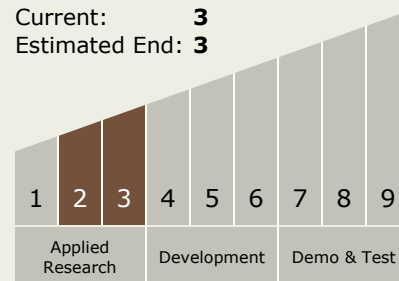
John R Lewis

## Technology Maturity (TRL)

Start: 2

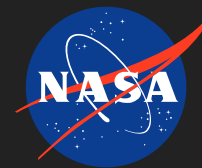
Current: 3

Estimated End: 3



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✓ **June 2016:** Closed out

**Closeout Summary:** Our exploration of the feasibility of in-space production of storable propellants from resources available on asteroids, and also on Mars and the Moon, has considered the process of sample acquisition; the energy requirements for heating and volatile release; the thermodynamic behavior of gas release from minerals and organic polymers containing hydrogen, oxygen, carbon, sulfur, and nitrogen; purification of the released water and carbon dioxide; the storage and transportation of these materials in the condensed state; synthesis of fuels including methanol and dimethyl ether (DME); and the co-production, concentration, stability, and storage of the complementary oxidizer high-test hydrogen peroxide (HTP). We call attention to the ability of the storable propellant/oxidizer combination of DME and HTP to carry out deep-space missions and to perform retrieval and relocation of any and all space-derived resources, such as retrieving asteroidal metal to high Earth orbit. This study's analyses are based on returning resources to a storage/processing/dispensing facility in a Highly Eccentric Earth Orbit (HEEO) with a perigee above geosynchronous orbit and an apogee approach or beyond the Moon's orbit. The methane/LOX option, notable for good engine performance, has not been included in this study because our scope includes only fully storable propellants. This work in Phase I has led to a number of conclusions. These are: 1. Thermodynamic theory shows that extraction of water and carbon dioxide from carbonaceous (C-type) near-Earth asteroids by means of direct solar heating is feasible and efficient. This is in agreement with experiments carried out by Joel Sercel in an independent NIAC Phase I project, using both CM type meteorite material and a C-type asteroid simulant that we have developed at DSI and the University of Central Florida under an SBIR grant running concurrently with this (and his) NIAC grant. 2. The same theory also predicts that attempts at full extraction of volatiles from a C asteroid will require calciner temperatures of at least 700 K, at which temperature not only does the native organic polymer in the C asteroid material react with magnetite ( $\text{Fe}_3\text{O}_4$ ) to generate carbon dioxide and water vapor, but also these gases react with coexisting sulfide and sulfate minerals to release copious amounts of sulfur dioxide. This prediction has also been qualitatively verified by Joel Sercel's work on meteorite and DSI simulant materials. At these temperatures, release of H, C, O, N, and S produces gases amounting to about 40% of the total mass of the asteroidal material. 3. The principal sulfur gas released, sulfur dioxide, is a source of some concern for several reasons. It is a toxic and offensively odorous gas that must be removed from water intended for life-support or hydroponic use. It also spontaneously generates elemental sulfur and sulfuric acid, highly undesirable materials that can corrode or clog water-handling systems such as Solar Thermal Propulsion engines. Further, some of these sulfur compounds can poison the catalyst beds used in several critical processing steps in manufacture of propellants and metal products. 4. Sulfur impurities may be removed on the asteroid from freshly generated impure water to make it safe and suitable for these uses. The application of Reverse Osmosis (RO) or other relevant technology provides the requisite purification in a simple and safe manner, using equipment of high TRL. The purified water is then suitable for use in everything from STP thrusters to chemical reagents to chemical propellant manufacture to life-support fluids. (4alt.) Alternatively,  $\text{SO}_2$  release may be minimized by using lower calcining temperatures, which also severely limits  $\text{CO}_2$  release. This option suggests retrieval of only water to HEEO on the first mission, with production of only H<sub>2</sub>O propellants. Such a mission would also provide an opportunity for retrieval of unprocessed asteroid material to HEEO for use in process-development experiments there, relegating  $\text{CO}_2$  and  $\text{SO}_2$  production to HEEO. 5. The rejected sulfur compounds from water purification are

## Technology Areas

### Primary:

- TX01 Propulsion Systems
  - ↳ TX01.1 Chemical Space Propulsion
    - ↳ TX01.1.2 Earth Storable

## Target Destinations

The Sun, Earth, The Moon, Mars, Outside the Solar System

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### Project Website:

<https://www.nasa.gov/directorates/spacetech/home/index.html>